

Fabrication of a porous ZnRh₂O₄ photocathode for photoelectrochemical water splitting under visible light irradiation and a significant effect of surface modification by ZnO necking treatment

著者	Kamimura Sunao, Higashi Masanobu, Abe Ryu, Ohno Teruhisa
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Electronic Supporting Information (ESI) for

Fabrication of a porous ZnRh₂O₄ photocathode for photoelectrochemical water splitting under visible light irradiation and significant effect of surface modification by ZnO necking treatment

Sunao Kamimura,^{1,2} Masanobu Higashi,³ Ryu Abe,³ and Teruhisa Ohno^{1,2,4*}

¹Department of Applied Chemistry, Faculty of Engineering, Kyushu Institute of Technology, 1-1 Sensuicho, Tobata, Kitakyushu 804-8550, Japan

² Research Center for Advanced Eco-fitting Technology, Kyushu Institute of Technology, 1-1 Sensuicho, Tobata, Kitakyushu 804-8550, Japan

³ Graduate School of Engineering, Kyoto University, Katsura, Nishikyo-ku, Kyoto 615-8510, Japan

⁴ ACT-C, Japan Science and Technology Agency, 4-1-8 Honcho, Kawaguchi-shi, Saitama 322-0012, Japan

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ESI. Fig. S1 XPS spectra of bare ZnRh₂O₄ and ZnO/ZnRh₂O₄ electrode.

ESI. Fig. S2 CV curve of bare ZnRh₂O₄ and ZnO/ZnRh₂O₄ electrode.

ESI. Fig. S3 XPS spectra of ZnO/ZnRh₂O₄ electrode before and after PEC reaction.

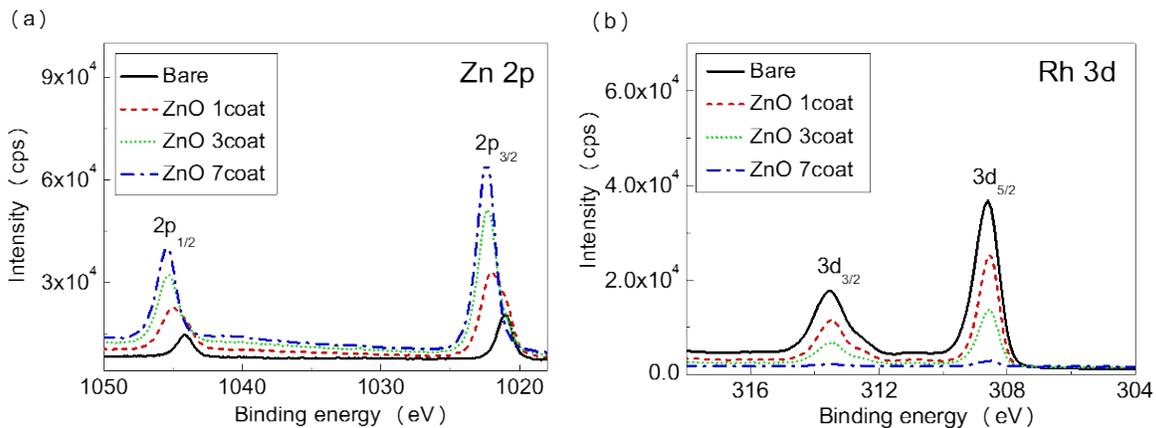


Figure S1. XPS spectra of bare ZnRh₂O₄ and ZnO/ZnRh₂O₄ electrode.

For the bare ZnRh₂O₄ (see **Fig. S1(a)**), major peaks at 1021.0 eV and 1044.1 eV were observed, which peaks were attributed to typical values of Zn 2p_{3/2} and 2p_{1/2} in ZnRh₂O₄, respectively. This our observation was similar to a previous report by Irie *et al.* (refer to *Journal of Materials Chemistry A*, 2016, **4**, 3061 – 3067). These Zn 2p XPS intensities were increased with an increase in the number of drop-casts of ZnO precursor solution, implying that amount of Zn species was increased by necking treatment. It should be noted that peak shift (~ ca.1 eV) of the Zn 2p XPS spectrum was observed after necking treatment; major peaks at 1022.4 eV and 1045.5 eV were newly appeared, and these peaks were attributed to typical values of Zn 2p_{3/2} and 2p_{1/2} in ZnO (refer to *Thin Solid Films* 2005, **491**, 153 – 160). In contrast, the Rh 3d XPS intensities were decreased with an increase in the number of drop-casts. These results suggest that ZnO amount at ZnO/ZnRh₂O₄ surface is increased with an increase in the number of drop-casts, which leads to inhibit of X-ray excitation of ZnRh₂O₄, resulting decrease of Rh 3d XPS intensity.

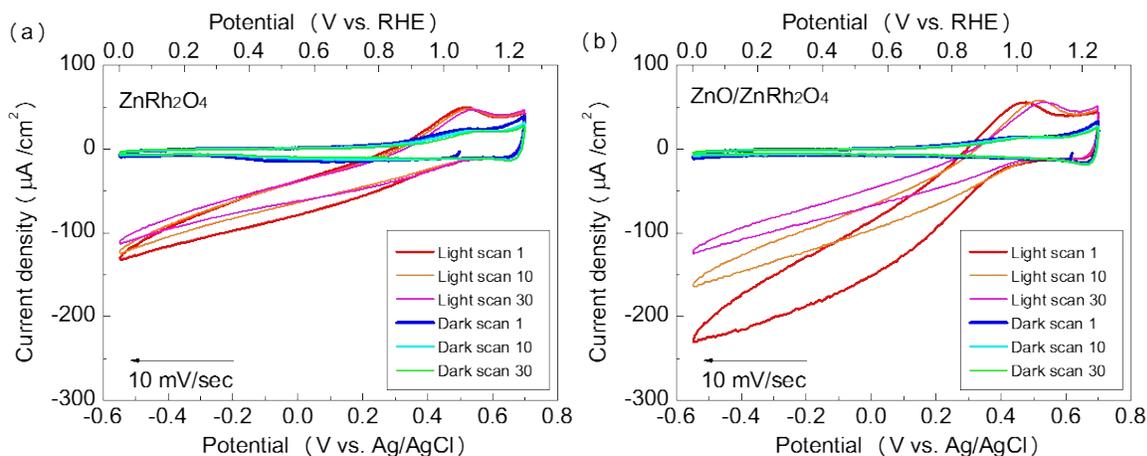


Figure S2. Cyclic voltammetry (CV) for bare and ZnO/ZnRh₂O₄ photocathodes.

The CV curves were measured for bare and ZnO/ZnRh₂O₄ photocathodes in Ar-purged 0.1 M Na₂SO₄ solution at potential sweep rate 10 mV/sec. Light source was used Xe-lamp equipped with L-42 cut off filter. The potential first to scan from rest potential to -0.6 V vs. Ag/AgCl at which the scan direction was reversed to +0.7 V vs. Ag/AgCl, and then scan back to the initial potential. It should be noted that the rest potential of bare and ZnO/ZnRh₂O₄ photocathodes were +0.50 V and +0.62 V vs. Ag/AgCl, respectively. As seen in this figure, the photocurrent response of bare ZnRh₂O₄ electrode was a relatively stable under PEC reaction. In contrast, the photocurrent of ZnO/ZnRh₂O₄ photocathode was gradually decreased with a number of cycles. This indicating that durability of ZnO/ZnRh₂O₄ photocathode is not good.

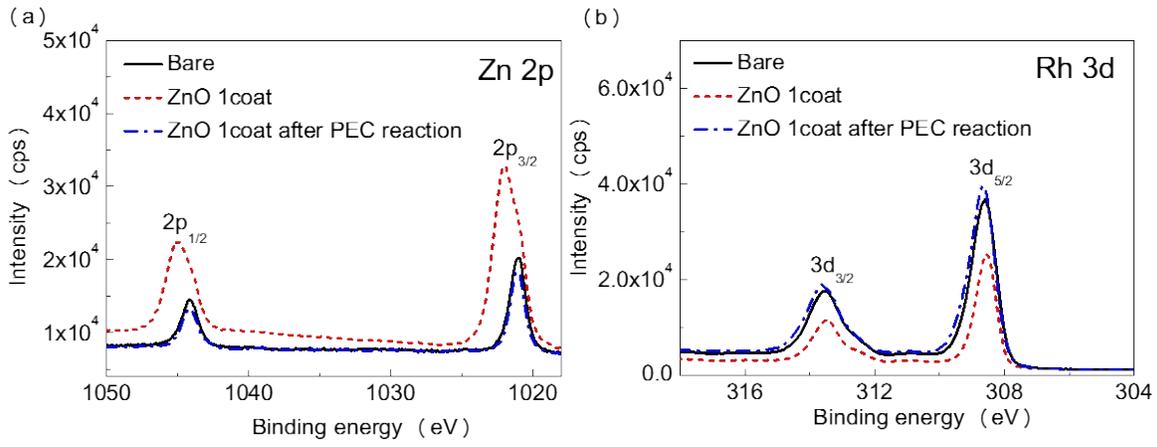


Figure S3. XPS spectra of ZnO/ZnRh₂O₄ electrode before and after PEC reaction.