1	First-principles study on hydrogen diffusivity in BCC, FCC, and HCP iron						
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18	Abstract						
19	The hydrogen diffusion behavior in BCC, FCC, and HCP iron has been investigated by means of first-principles						
20	calculations. Diffusion coefficients were estimated quantitatively from the migration energy derived by the Nudged						
21	elastic band method, and phonon calculations including the vibrations of all atoms at every stable and metastable site.						
22	Our calculations on the BCC structure show good agreement with those in the previous report. In the FCC structure						
23	as well, the calculated diffusion coefficients are in good agreement with experimental data. Our results suggest that						
24	the consideration of the antiferromagnetic state in FCC is important for the reproduction of experimental results. For						
25	the HCP structure, although there was a lack of systematic experimental results, our calculations predict that the						
26	diffusion coefficient is smaller than that in the case of the FCC sample. In the HCP lattice, there are two diffusion						
27	paths: one parallel to the c-axis and the other in the c-plane. The direction and the diffusion coefficient can be						
28	controlled by the tuning of c/a , which is the ratio of the lattice constants.						

32 **1. Introduction**

Hydrogen is being increasingly used as a clean energy source that reduces the amount of carbon dioxide emitted 33 into the atmosphere. Therefore, hydrogen-resistant structure materials have been sought for hydrogen-energy-related 3435infrastructure. In this regard, steels are advantageous with respect to the strength and cost performance. Key factors 36 affecting mechanical performance of steels under hydrogen environment are hydrogen content, distribution, and 37 diffusivity. In particular, the diffusivity of hydrogen drastically alters fatigue properties which are the determinant 38factor for safety structure design. From a viewpoint of hydrogen diffusivity, close-packed structures are believed to show a low diffusivity of hydrogen, i.e. FCC and HCP structures. In steels, austenitic steels actually show a low 39 40 diffusivity [1-3], which results in high resistance to fatigue crack growth in hydrogen environments [4].

On the other hand, there is few systematic knowledge about hydrogen diffusivity for another promising candidate, 4142HCP structure. Because stable and single HCP phase cannot be obtained under ordinary pressure. Instead, deformation-induced HCP-martensitic transformation from austenitic phase was utilized to obtain high resistance to 4344fatigue crack growth in Fe-Mn-Cr-Ni steel, which successfully endowed comparable fatigue crack growth resistance to stable austenitic steels [5]. Correspondingly, some indirect evidences of low diffusivity of HCP-martensite has 4546been reported [6,7]. The utilization of HCP martensite is expected to give a new pathway for alloy design of hydrogen-resistant steels. However, because the effect of lattice defects including its interface on diffusivity cannot 4748be quantified, experimental methodologies such as permeation test cannot determine underlying diffusivity along 49 interstitial sites of HCP. Furthermore, the c/a and volume of HCP change sensitively by composition of alloys and 50constraint condition with parent phase. Thus, designing HCP-based hydrogen-resistant materials needs systematic 51and atomic-level insight revealed from the computational simulations.

52From a theoretical point of view, the diffusion behavior of hydrogen has been studied using several simulation 53techniques [8-13]. First-principles calculations based on the density functional theory (DFT) have been used to 54determine the energetics associated with the migration of interstitial H atoms in crystalline metals. Since hydrogen is the lightest chemical element, its diffusion is affected by quantum mechanical effects. Quantum mechanical effects 5556on the diffusion of interstitial atoms can be divided into two contributions. The first is due to the zero point energy 57(ZPE) of the ground state of a vibrating proton or, more generally, due to the presence of discrete vibrational energy levels and their occupied state at finite temperatures. The second contribution is due to quantum mechanical tunneling. 5859Kimizuka et al.[9] and Stefano et al.[10] have pointed out that it is difficult to explain the diffusivity of hydrogen 60 with accuracy in BCC materials using the transition state theory (TST) since it cannot account for the quantum effect 61 become prominent at low temperature. On the other hand, the diffusion coefficient calculated using the TST is in 62good agreement with experimental results for Ni and Ti that have the FCC and HCP closed packed structures, 63 respectively[10,12]. This observation could be due to the quantum effect being negligible in those metals. Therefore, 64 an investigation based on the TST in FCC and HCP irons would be insightful. Comparison of the hydrogen diffusivity 65 in BCC, FCC, and HCP, missing piece of the study, make a significant contribution toward materials design for 66 hydrogen-resistant steels.

In this study, we have investigated the hydrogen diffusion behavior in BCC, FCC, and HCP iron by means of firstprinciples calculations. Our calculation results for the BCC iron show good agreement with those of previous reports [9,10]. In the FCC iron, the calculated diffusion coefficients are compared with the experimental data. Our

- calculations predict that the diffusion coefficient of the HCP sample which has several c/a is smaller than that of the
- FCC sample. In the HCP lattice, there are two diffusion paths: one parallel to the *c*-axis and the other in the *c*-plane.
- The direction and the diffusion coefficient can be controlled by the tuning of c/a, which is the ratio of the lattice
- constants. The low and direction controllability of hydrogen diffusivity in HCP iron suggests a different paradigm in
- steels under hydrogen environment.

75 **2. Computational details**

Following Wert and Zener [14] the interstitial diffusion coefficient is described as

$$77 \qquad D = nL^2\Gamma \qquad (1)$$

where *n* is a numerical coefficient, whose value depends on the location of the interstitial positions, *L* and Γ are the

jump distance projected onto the diffusion direction and the jump rate, respectively. In the Kehr approximation [15],
 the jump rate can be written as follows:

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$$\Gamma = \frac{k_{\rm B}T}{h} \exp(-\frac{\Delta E}{k_{\rm B}T})$$
 (2)

where ΔE is the migration energy in the diffusion pathway. k_B, T and h indicate the Bolzmann constant, temperature and Planck constant, respectively. The simplest extension of this classical expression taking into account zero-point energy, the temperature-dependent part of the internal energy and the vibrational entropy is usually done by simply adding the difference in vibrational free energies, ΔG_{vib} , corresponding to the saddle point and the initial state, to the migration energy in the exponential term in Eq.(2). The corrected diffusivity is hence expressed as

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$$D = nL^2 \frac{k_B T}{h} \exp(-\frac{\Delta E + \Delta G_{\text{vib}}}{k_B T})$$
(3)

88 where

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$$\Delta G_{\rm vib} = G_{\rm vib}^{\rm saddle} - G_{\rm vib}^{\rm interstitial site}$$
 (4)

90 The vibrational free energy G_{vib} in several states is explained as

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$$G_{\rm vib} = -k_{\rm B}T \int_0^\infty g(\upsilon) \ln[2\sinh(\frac{h\upsilon}{2k_{\rm B}T})]d\upsilon \qquad (5)$$

92 where g(v) is the phonon density of states and v is vibrational frequency.

In order to calculate ΔE for hydrogen diffusion in iron, the climbing image nudged elastic band method[16] (CI-93 NEB) was used. The NEB method provides the means to find a minimum energy path (MEP) when the initial and 9495 final states of a process are known [17]. The CI-NEB method is an improved version of the NEB method, and it is used to locate the MEP and transition state for the diffusion of hydrogen in bulk Fe. We used the VTST code for the 96 Vienna Ab initio Simulation Package (VASP) code [18,19] developed by Henkelman et al. [16]. The supercells of 97 Fe₅₄H ($3 \times 3 \times 3$), Fe₃₆H ($2 \times 2 \times 2$), and Fe₉₆H ($4 \times 4 \times 3$) were used for BCC, FCC, and HCP, respectively in the 98 CI-NEB calculation. We set linear interpolation images between the initial and final states with three or five images. 99 100 which were converged to the MEP with a $3 \times 3 \times 3$ mesh for k-point sampling in BCC and HCP. We used a $5 \times 5 \times$

101 5 mesh for k-point sampling with respect to FCC.

102 The lattice relaxation and total energy calculations were performed with the VASP code, which is based on DFT. 103 The exchange and correlation functions were given by the generalized gradient approximation, as proposed by 104 Perdew *et al.*[20]. We employed Blochl's projector-augmented wave (PAW) method as implemented by Kresse and 105 Joubert [21, 22]. The Monkhorst-Pack method [23] is used to set the k-point mesh and a cutoff of 334.9 eV for the 106 plane waves was applied. Before the NEB calculations were performed, structure optimizations were performed with 107 a $13 \times 13 \times 13$ mesh for k-point sampling with a cutoff energy of 348.247 eV regarding BCC and FCC. In the 108 calculation of HCP, we used a $15 \times 15 \times 9$ mesh for k-point sampling.

- 109In this study, the magnetic properties were assumed to be ferromagnetic in BCC, antiferromagnetic in FCC, and nonmagnetic in HCP samples. In the FCC iron, magnetic states have been investigated and it was suggested that the 110antiferromagnetic state was stable [24-26]. However, the double-layered antiferromagnetic structure [25] causes the 111 lattice symmetry to change from cubic to tetragonal. Therefore, these two lattice types, namely, antiferromagnetic 112113FCC and antiferromagnetic FCT, were generated in the following manner. First, the lattice parameters were optimized 114in the nonmagnetic FCC and HCP. The optimized lattice constants were a = 3.44 Å in FCC and a = 2.46 and c = 3.89115Å in HCP, respectively. These values are consistent with those reported in previous works[8,10,24,26]. The FCC structure, which was optimized for the antiferromagnetic state, resulted in a tetragonal structure, which had a = 3.46116 117and c = 3.75 Å. The difference in the equilibrium volume between the antiferromagnetic FCT and nonmagnetic FCC is estimated to be 10.3 %, which is expected to affect the diffusivity of hydrogen. Hence, we calculated the migration 118energy of the antiferromagnetic FCT and FCC, which have the same equilibrium volume, by the NEB method. On 119 120the other hand, although HCP has been calculated as antiferromagnetic state[26], the difference in the equilibrium volume between the antiferromagnetic and nonmagnetic state was only 2.7%, which is comparatively lower than that 121122of FCC. Therefore, we have calculated the migration energy of hydrogen in the HCP in the nonmagnetic state since 123the effect of the volume change is expected to be low.
- 124 Thermodynamic functions are obtained from phonon dispersions, which are calculated by the direct method [27] 125 with ab initio forces from VASP. To estimate the force due to displacement within the DFT calculation, we used the 126 plane wave energy cut off of 267.882eV, the convergence criterion for electric self-consistency loop of 10^{-6} eV, a Γ 127 point centered k-mesh of $2 \times 2 \times 2$, the Methefessel-Paxton smearing method with the width of the smearing in 0.2 128 eV, and a displacement of 0.01 Å.
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147 **3. Results and discussion**

148 **3.1. BCC**

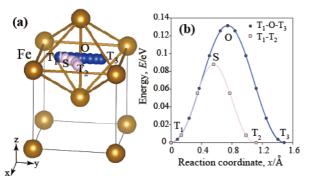




Fig.1. (a) The hydrogen migration pathways in BCC. O, T, and S represent the octahedral site, tetrahedral site, and saddle point, respectively. (b) The energy profiles for two paths: T_1 - T_2 direct and T_1 -O- T_3 indirect.

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First, we have examined our calculated results for BCC and compared them with those reported in previous 153154studies[8,10]. Figs.1 (a) and (b) show the migration pathways of the hydrogen atom in the BCC and the energy profiles for T₁-T₂ direct and T₁-O-T₃ indirect hopping, respectively. The minimum energy path along T₁-O-T₃ is 155almost straight, while along T1-T2 it seems to curve towards the O-site. This behavior is consistent with those reported 156in a previous paper [8]. From Fig. 1(b), it is revealed that the T_1 - T_2 direct hopping is a more stable transition path than 157T₁-O-T₃ indirect hopping. The "bare" migration energy ΔE can be estimated to be 0.088 eV. The calculated phonon 158159dispersions of several states are given in Figs. 2. In fig.2(a), two branches at 29.6 and 43.7 THz correspond to the 160vibration of the hydrogen atom at tetrahedral interstitial site. The branches of hydrogen vibration derived from 161asymmetry of BCC lattice with hydrogen in T-site. The results have good agreement with the calculated values reported by Sakagami et al. [28]. From the Fig.2(b) we see that the hydrogen has one imaginary mode of vibrational 162frequency at the transition state of T_1 - T_2 . The imaginary mode means that there is dynamical instability at the site. 163164The stable transition state for the diffusion has only one imaginary mode. Therefore, the S-site is a stable transition 165state. Furthermore, we estimated that the ΔZPE was 0.053 eV from these energy frequencies. The ZPE-corrected migration energy was 0.035 eV, which is in agreement with 0.042eV and 0.044eV presented in previous reports[8,10]. 166167

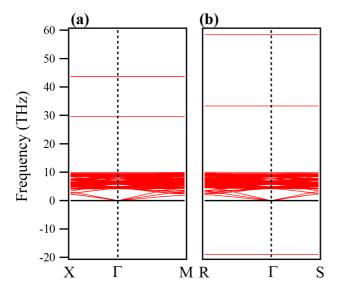
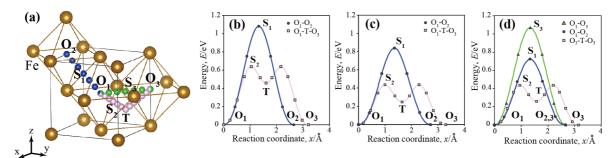


Fig. 2. The calculated phonon dispersions of a 56 atom-Fe supercell with hydrogen atom in (a) T-site and (b) the saddle (S) point of T_1 - T_2 , respectively. The symmetry of the supercell describing the transition state is different from that of the tetrahedral site, which is reflected in the labeling of the special points in the Brillouin zone.

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175 **3.2. FCC**

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Fig.3. (a) The hydrogen migration pathways in FCC. (b) The energy profiles for O-T-O indirect and O-O direct paths in nonmagnetic, (c) antiferromagnetic FCC, and (d) antiferromagnetic FCT, where the O_1-O_2 , and O_1-O_3 paths are distinguished.

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182Fig. 3(a) shows the migration pathways in the FCC and FCT lattices considered in this study. In the FCC case, the 183stable interstitial site is the O-site, which is consistent with the result of Ismer et al.[11]. Three types of migration pathways between the O-sites are indicated, namely, O1-O2 direct, O1-T-O3 indirect, and O1-O3 direct (which is 184identical to the O_1 - O_2 path in the case of FCC) paths. The energy profiles for these migration pathways in 185186nonmagnetic FCC, antiferromagnetic FCC, and antiferromagnetic FCT are shown Fig.3 (b)-(d). In antiferromagnetic 187 FCT, the two paths of the direct nearest neighbor O-sites are distinguished. In all cases, the energy barrier between 188 the nearest neighbor O-sites mediated by the T-site are lower than O_1-O_2 and O_1-O_3 direct hopping. Therefore, the 189 O1-T-O3 indirect migration pathway can be considered as a stable one. For antiferromagnetic FCT, there are two O-

- T-O paths. One is in the same spin layer and another is the path between different spin layers. The more stable path is found in the same spin layer and the migration energies of all the diffusion pathways are listed in Table 1. The energy barrier of the antiferromagnetic FCC (0.44 eV) and FCT (0.44 eV) are consistent with the experimental migration energy, which was reported as 0.46 eV[29].
- 194 Figure 4 shows the phonon dispersion for each atomic configurations, (a) nonmagnetic FCC supercell with hydrogen
- 195 in O-site and T-site, (b) saddle point, (c) antiferromagnetic FCC supercell with hydrogen in the interstitial sites in O-
- 196 site and T-site, (d) saddle point, and (e) antiferromagnetic FCT. In Fig. 4(a), (c), and (e), several dispersionless mode
- 197 mainly originated from the hydrogen vibration for different configurations are shown in the same figure, because the
- phonon dispersion derived from host Fe atoms is slightly changed. The hydrogen vibrations in an FCC Fe-25Cr-20Ni stainless steel alloy with a low hydrogen content has been reported by neutron spectroscopy [30]. The vibrational energy has been found to be \sim 130 meV(31 THz). This value has consistent with calculated vibrational energies of nonmagnetic FCC (139 meV), antiferromagnetic FCC (140 meV) and antiferromagnetic FCT (137 meV), respectively.
- 203For the nonmagnetic FCC with hydrogen in O-site and T-site, there is no imaginary mode. Therefore we would like 204to point out that the T-site is not unstable, but a metastable site. On the other hand, the state with hydrogen in saddle point has one imaginary mode. Thus, the stable diffusion pathway of hydrogen is thought to be between the O-sites 205mediated by the T-sites in FCC. A similar behavior observed in antiferromagnetic FCC and FCT is shown in fig. 4 206207(c) \sim (e). From the figures, O-site and T-site are indicated as a stable interstitial site, and the saddle point between O1-T in antiferromagnetic FCC and FCT have one imaginary mode and these states are revealed stable transition 208209 state of hydrogen diffusion. Ismer et al[11]. have reported similar results from first-principles calculations. Recently, Machida et al.[31] have deduced that the interstitial site for hydrogen in FCC is not only the O-site but also the T-210211site from neutron diffraction patterns under high pressure. Thus, our results are consistent with the calculated and 212experimental results.
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Table 1 The calculated migration energy for hydrogen diffusion in nonmagnetic FCC, antiferromagnetic FCT and FCC, respectively.

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218		Nonmagnetic	Antiferromagnetic	Antiferromagnetic
		FCC (eV)	FCT (eV)	FCC (eV)
219	O ₁ -T-O ₃	0.64	0.44	0.44
220	0]-1-03	0.04	0.44	0.44
221	O_1 - O_3	1.08	1.07	0.84
222	O_1 - O_2	-	0.72	-
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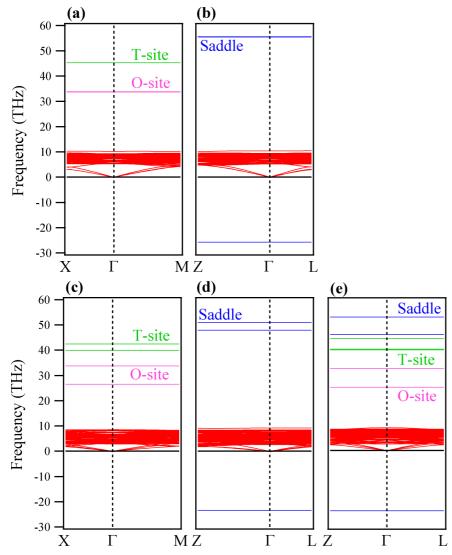


Fig. 4. The calculated phonon dispersions of a 32-atom Fe supercell with hydrogen atom in (a) O-site and T-site, and (b) saddle point in nonmagnetic FCC, respectively. The calculated phonon dispersions of antiferromagnetic FCC with hydrogen which states in (c) O-site and T-site, (d) saddle point, respectively. (e) the calculated phonon dispersions of antiferromagnetic FCT supercell with hydrogen which occupies several sites. The labeling of the special points in the Brillouin zone depends on the symmetry of several supercells. The pink, green and blue lines indicate the vibrational frequencies of hydrogen in O-site, T-site and saddle point, respectively.

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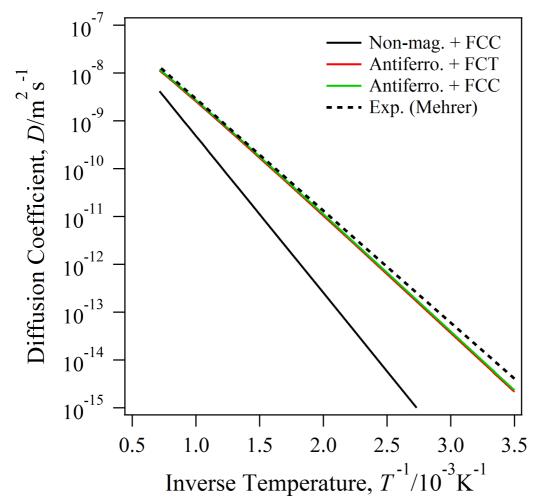


Fig.5. The diffusion coefficient of hydrogen in FCC. The calculated values of nonmagnetic FCC, and
antiferromagnetic FCT and FCC are denoted by the solid, red, and green lines, respectively. The dashed line
indicates the experimental diffusion coefficient by Mehrer[29].

236

241 Diffusion coefficient of hydrogen between nearest stable O-site in FCC are described as

$$242 \qquad D_{FCC} = a^2 \Gamma \qquad (6)$$

where *a* is a lattice constant. In this case, however, hydrogen diffusion occurs via metastable T-site. Therefore diffusion coefficient is modified as follows,

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$$D'_{FCC} = \frac{1}{2} D_{FCC} (1 + 2 \exp(-\frac{\Delta G_{\text{tet-oct}}}{k_{\text{B}}T}))^{-1}$$
 (7)

 $\Delta G_{\text{tet-oct}}$ is the difference of the free energy between hydrogen in tetrahedral and octahedral sites. Fig.5 shows an Arrhenius plot of the diffusion coefficient of hydrogen in nonmagnetic FCC, and antiferromagnetic FCC and FCT along with the experimental diffusion coefficient in pure FCC between 1183K (1000/T = 0.845) and 1373K (1000/T = 0.728) [29]. Our calculation results of the nonmagnetic state agree with those presented in a previous report[11]. The calculated diffusion coefficients of antiferromagnetic FCC and FCT are approximately same, and closely agree with the experimental value for pure Fe. These results are consistent with a fact that the FCC has the antiferromagneticstate as the ground state.

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255 3.3. HCP

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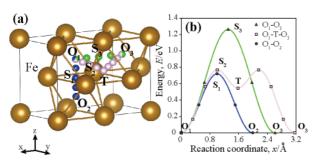


Fig.6. (a) The hydrogen migration pathways in HCP. (b) The energy profiles for several paths in the case of c/a = 1.58.

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Fig. 6(a) shows the hydrogen atoms on migration pathways in the HCP iron. In this study, a fully relaxed lattice was obtained with c/a = 1.58. The migration energies of several diffusion paths are shown in Fig. 6(b). The stable interstitial site is found to be the O-site, while the T-site is metastable. The migration energies of O₁-O₂ direct, O₁-T-O₃ indirect, and O₁-O₃ direct hopping are estimated to be 0.72, 0.77, and 1.26, respectively. In case of c/a = 1.58, it is revealed that the O₁-O₂ direct diffusion path along the *c*-axis is slightly more stable than the diffusion path via the T-site on *c*-plane.

The calculated phonon dispersion of nonmagnetic HCP supercell with hydrogen in O-site and T-site is shown in Fig. 7 (a). Fig. 7 (b) and (c) show the calculated phonon dispersions of the saddle points S_1 between O_1 - O_2 and S_2 between O_1 -T. The hydrogen vibrational energy of dhcp FeH has been reported as 103meV by inelastic neutron scattering spectrum [32] and calculated value (140 meV, 110meV) with hydrogen in O-site is slightly higher than the experimental value. The difference of vibrational energy between experimental and calculated values can be caused by the difference of the metal-hydrogen distance.

For the state with hydrogen in O-site and T-site, there no imaginary mode. Therefore, these interstitial sites are stable from the view point of dynamics. On the other hand, the two states of saddle point have one imaginary mode derived from hydrogen. Therefore, both the O_1 - O_2 path along the *c*-axis and the O_1 -T- O_3 path in the *c*-plane can serve as stable diffusion paths in HCP.

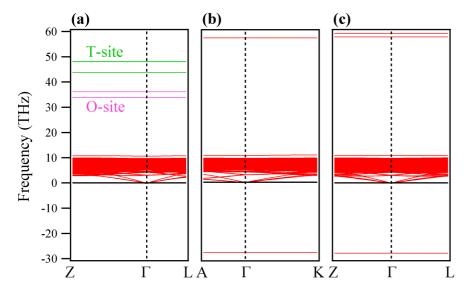


Fig. 7. The calculated phonon dispersions of a 96-atom Fe supercell with hydrogen atom in (a) O-site and T-site,

(b) the saddle point of S_1 between O_1 - O_2 and (c) S_2 between O_1 -T in nonmagnetic HCP, respectively. The labeling of the special points in the Brillouin zone depends on the symmetry of several supercells. The pink and green lines indicate the vibrational frequencies of hydrogen in O-site and T-site, respectively.

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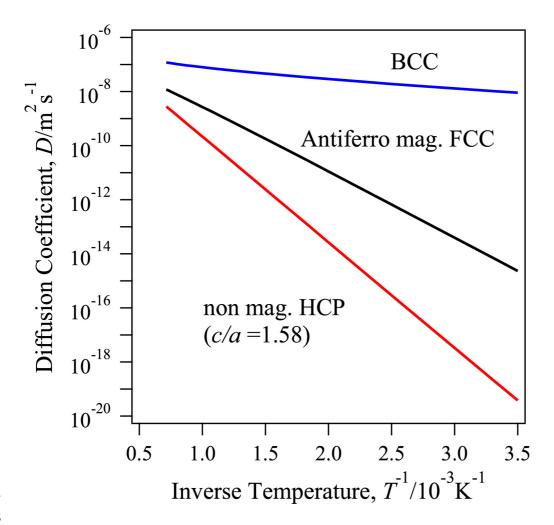


Fig.8 The Arrhenius plot of the diffusion coefficient of hydrogen in HCP and antiferromagnetic FCC. The red solid line indicates the diffusion coefficient for HCP with c/a = 1.58. The black and blue lines indicate the diffusion coefficient for antiferromagnetic FCC and BCC, respectively.

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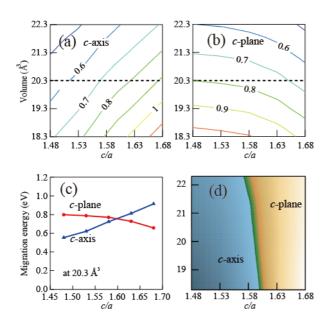
Here the diffusion coefficient in BCC, FCC, and HCP are compared. Although the hydrogen diffusion in BCC is isotropic, that in HCP is anisotropic along *c*-axis. Therefore, hydrogen diffusion in HCP must be considered with a coordination number between nearest stable O-site along *c*-axis. The diffusion coefficient in BCC and HCP are described as follows, respectively.

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$$D_{BCC} = \frac{1}{12}a^2\Gamma$$
 (8)

$$298 \qquad D_{HCP} = \frac{1}{4}c^2\Gamma \qquad (9)$$

Fig.8 shows an Arrhenius plot of the diffusion coefficient of hydrogen in HCP for c/a = 1.58, along with relevant data of the antiferromagnetic FCC and BCC. The diffusion coefficient for HCP is revealed to be lower than that for FCC and BCC. Hence, materials that have the HCP structure can be considered to have the potential for resistance against hydrogen embrittlement that could either equal or even surpass FCC materials. This result has qualitatively agreement with the experimental result: the hydrogen diffusion behavior has not been changed on epsilon martensitic transformation, which is the transformation from FCC to HCP, by Koyama *et al.* [6, 7]. On the other hand, c/a of FeMn steels with epsilon martensite was reported to be 1.63 from experiments[33]. The different c/a values obtained from different experiments are assumed to be due to the additive elements present in the alloy. Therefore, we have investigated the c/a and volume dependence of hydrogen diffusivity in HCP because these values are expected to be influenced by additive elements in actual materials.

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Fig.9. The migration energy in eV of (a) O_1 - O_2 along *c*-axis and (b) O_1 -T- O_3 on the *c*-plane with respect to volume and *c/a* of HCP. (c) Comparison of the hydrogen migration energies along *c*-axis and on the *c*-plane on the crosssection, denoted by dashed line in (a) and (b). (d) The difference of migration energy between *c*-axis and *c*-plane. The left and right sides indicates *c*-axis and *c*-plane diffusion dominated region, respectively.

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From the calculated energy profiles, we found an interesting behavior of hydrogen in HCP iron. This was the anisotropic diffusion of hydrogen depending on c/a. Fig.9 shows the relationship between the migration energy, volume, and c/a of HCP. The migration energies of O₁-O₂ along the *c*-axis and O₁-T-O₃ on the *c*-plane are indicated in Fig.9(a) and (b), respectively. The equilibrium volume and c/a of nonmagnetic HCP are 20.3 Å³ and 1.58, respectively. The level lines of migration energy of O₁-O₂ are at about 45° and are assumed to be affected equally by a change in volume and c/a ratio. On the other hand, the lines of migration energy of O₁-T-O₃ seem to shift slightly with changing c/a ratio and hence they are considered to be affected to a lower extent by the change in c/a.

The *c/a*-dependent migration energies of both the pathways are shown in Fig.9(c). The data corresponds to the crosssection marked using dashed lines in Fig.9(a) and (b). The diffusion pathway in the *c*-plane becomes stable with increasing *c/a* values whereas the path along the *c*-axis becomes unstable. The stable pathways are found to be along the *c*-axis below $c/a \sim 1.6$ and in the *c*-plane above $c/a \sim 1.6$. Fig. 9(d) indicates the stable pathways with respect to volume and c/a. The blue and orange regions indicate which is the stable diffusion paths. It is revealed that the stable diffusion path is along the *c*-axis O₁-O₂ below $c/a \sim 1.6$ and the *c*-plane O₁-T-O₃ becomes stable above $c/a \sim 1.6$.



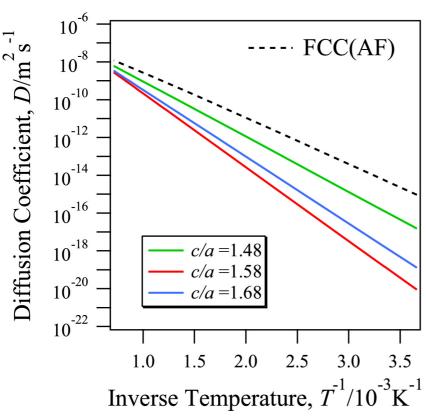


Fig.10. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP with c/a = 1.48, 1.58, and 1.68, respectively. Dashed line indicates the calculated results of antiferromagnetic FCC.

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From the migration energies investigated above, the diffusion coefficient was estimated in the range of c/a = 1.48– 1.68. For the calculation of diffusion coefficient in the case of c/a = 1.68, the diffusion path contains the metastable T-site. Therefore diffusion coefficient is modified as follows,

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$$D'_{HCP} = \frac{3}{2} D_{c-plane} (1 + 2 \exp(-\frac{\Delta G_{\text{tet-oct}}}{k_B T}))^{-1}$$
 (10)

340 where

341
$$D_{c-plane} = \frac{3}{2}a^2\Gamma$$
 (11).

The equation (11) was developed by considering a coordination number between stable O-sites in *c*-plane. The 342Arrhenius plots of the diffusion coefficient of hydrogen in HCP at 20.3 Å³ are given in Fig.10. Minimum diffusivity 343 is observed around $c/a \sim 1.6$, and the diffusion coefficient increases with both increasing and decreasing c/a ratio. 344However, the direction of the H diffusion changes with the value of c/a. When c/a is higher than ~1.6, the diffusion 345346path is in the *c*-plane O_1 -T- O_3 . On the other hand, when c/a is lower than ~1.6, the diffusion path is along the *c*-axis. 347Recently, the diffusion coefficient of hydrogen in HCP has been reported by theoretical calculation for $c/a \sim 1.58$ 348 [13]. On the other hand, actual experimental c/a in FeMn steels was estimated at ~ 1.63[33]. Our findings regarding 349 the diffusivity in the HCP lattice has important implications in terms of the possibility of control over the magnitude and direction of the diffusion behavior. By tuning the c/a ratio, which can be performed by adding some elements into the iron, the magnitude and direction of the H diffusion can be controlled independently in the range c/a < 1.6or c/a > 1/6. For example, the c/a ratio of HCP-Fe can be changed by loading high pressure [34] and addition of alloying elements [35]. In fact, the change with respect to the c/a ratio of HCP-Co bulk and film has been reported by these methods[36,37]. This diffusion characteristic, which is a general physical property of the HCP lattice, can now be used to produce a new functionality of the HCP alloy system.

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361 **4. Conclusion**

362Hydrogen diffusivity in HCP iron, which is the promising candidate for high resistance to fatigue crack growth has been investigated by a theoretical calculation and compared to those of BCC and FCC. In BCC and FCC iron, the 363 364 calculated results with respect to stable interstitial sites are in good agreement with those reported in previous experimental and theoretical investigations. In the FCC case, the stable diffusion path is between the O-sites mediated 365366 by the T-site. The calculated hydrogen diffusion coefficient was enhanced and agreed with experimental results, 367 when the antiferromagnetic state was considered. In HCP, the interstitial site was found to be the O-site and the migration energy for the hydrogen diffusion was estimated to be higher than that in FCC, which causes low diffusivity 368 369 of H in the HCP lattice. The hydrogen diffusion was seen to have an anisotropic behavior that depended on the c/a370ratio. At an equilibrium volume of nonmagnetic HCP, the stable pathways were found to be the direct O-site path 371along the *c*-axis when c/a < 1.6 and the path between the O-sites mediated by the T-site on the *c*-plane when c/a > 1.61.6. The diffusion coefficient of hydrogen in HCP was found to be lower than that in FCC and it can be concluded 372that the epsilon martensite has a low susceptibility to hydrogen embrittlement. The low and direction controllability 373of hydrogen diffusivity in HCP iron suggests a different paradigm in steels under hydrogen environment. 374

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439	List of figure captions
440	
441	Figure 1. (a) The hydrogen migration pathways in BCC. O, T, and S represent the octahedral site, tetrahedral site,
442	and saddle point, respectively. (b) The energy profiles for two paths: T ₁ -T ₂ direct and T ₁ -O-T ₃ indirect.
443	6
444	
445	Figure 2. The calculated phonon dispersions of a 56 atom-Fe supercell with hydrogen atom in (a) T-site and (b) the
446	saddle (S) point of T ₁ -T ₂ , respectively. The symmetry of the supercell describing the transition state is
447	different from that of the tetrahedral site, which is reflected in the labeling of the special points in the
448	Brillouin zone.
449	7
450	
451	Figure 3. (a) The hydrogen migration pathways in FCC. (b) The energy profiles for O-T-O indirect and O-O direct
452	paths in nonmagnetic, (c) antiferromagnetic FCC, and (d) antiferromagnetic FCT, where the O ₁ -O ₂ , and
453	O ₁ -O ₃ paths are distinguished.
454	7
455	
456	Figure 4. The calculated phonon dispersions of a 32-atom Fe supercell with hydrogen atom in (a) O-site and T-site,
457	and (b) saddle point in nonmagnetic FCC, respectively. The calculated phonon dispersions of
458	antiferromagnetic FCC with hydrogen which states in (c) O-site and T-site, (d) saddle point, respectively.
459	(e) the calculated phonon dispersions of antiferromagnetic FCT supercell with hydrogen which occupies
460	several sites. The labeling of the special points in the Brillouin zone depends on the symmetry of several
461	supercells. The pink, green and blue lines indicate the vibrational frequencies of hydrogen in O-site, T-site
462	and saddle point, respectively.
463	9
464	
465	Figure 5. The diffusion coefficient of hydrogen in FCC. The calculated values of nonmagnetic FCC, and
466	antiferromagnetic FCT and FCC are denoted by the solid, red, and green lines, respectively. The dashed
467	line indicates the experimental diffusion coefficient by Mehrer[29].
468	10
469	
470	Figure 6. (a) The hydrogen migration pathways in HCP. (b) The energy profiles for several paths in the case of $c/a =$
471	1.58.
472	11
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474	Figure 7. The calculated phonon dispersions of a 96-atom Fe supercell with hydrogen atom in (a) O-site and T-site,
475	(b) the saddle point of S_1 between O_1 - O_2 and (c) S_2 between O_1 -T in nonmagnetic HCP, respectively. The
476	labeling of the special points in the Brillouin zone depends on the symmetry of several supercells. The

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pink and green lines indicate the vibrational frequencies of hydrogen in O-site and T-site, respectively. ...12

480Figure 8. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP and antiferromagnetic FCC. The red481solid line indicates the diffusion coefficient for HCP with c/a = 1.58. The black and blue lines indicate the482diffusion coefficient for antiferromagnetic FCC and BCC, respectively.

- Figure 9. The migration energy in eV of (a) O₁-O₂ along *c*-axis and (b) O₁-T-O₃ on the *c*-plane with respect to volume
 and *c/a* of HCP. (c) Comparison of the hydrogen migration energies along *c*-axis and on the *c*-plane on the
 cross-section, denoted by dashed line in (a) and (b). (d) The difference of migration energy between *c*-axis
 and *c*-plane. The left and right sides indicates *c*-axis and *c*-plane diffusion dominated region, respectively.
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- 491Figure 10. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP with c/a = 1.48, 1.58, and 1.68,492respectively. Dashed line indicates the calculated results of antiferromagnetic FCC.
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Table 1 The calculated migration energy for hydrogen diffusion in nonmagnetic FCC, antiferromagnetic FCT andFCC, respectively.

497				
498		Nonmagnetic	Antiferromagnetic	Antiferromagnetic
499		FCC (eV)	FCT (eV)	FCC (eV)
	O ₁ -T-O ₃	0.64	0.44	0.44
500	O ₁ -O ₃	1.08	1.07	0.84
501		1.00		0.04
502	O ₁ -O ₂	-	0.72	-